Investigation of Caffeic Acid Effect on Human Cancer Cell Line and Some Enzymes

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Abstract

Caffeic acid, a prominent antioxidant compound, has garnered significant attention in research due to its multifaceted properties, which include antiinflammatory, antiobesogenic, antithrombotic. vasodilating, and antitumor activities. This study aims to comprehensively investigate the impact of caffeic acid on various metabolic enzymes (carbonic anhydrase I, II, IX, and glutathione reductase) through both in silico and in vitro approaches. Furthermore, in vitro experiments did conducted on the AGS (gastric cancer cell) line and the HaCaT (keratinocyte normal cell) line to elucidate the effects of caffeic acid in these cellular systems.

Keywords: Caffeic acid, carbonic anhydrase, glutathione reductase, cancer.

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1. Introduction

Caffeic acid (CAA) emerges as a polyphenolic entity synthesized via the secondary metabolic pathways occurring within a diverse range of vegetables, notably encompassing olives, coffee beans, fruits, potatoes, carrots, and propolis. Moreover, this constituent assumes a pivotal role as the primary hydroxycinnamic acid derivative prevalent in the human dietary regimen (SILVA et al., 2014; TOSOVIC, 2017).

This phenolic compound (CAA), exists in a range of molecular configurations within the plant kingdom. It is encountered in its elemental state as monomers, taking the form of sugar esters, organic acid esters, glycosides, and amides. Additionally, CAA can adopt more intricate structures, such as dimers, trimers, and flavonoid derivatives. It is also capable of forming associations with proteins and other polymers located in the cellular wall of vegetables (CHEN and HO, 1997; VERMA and HANSCH, 2004; ESPÍNDOLA et al., 2019; SENTURK et al., 2022).

CAA assumes a crucial role in the defense mechanisms deployed by plants to fend off predators, pests, and infections. This compound exhibits inhibitory properties against the proliferation of insects, fungi, and bacteria (TOSOVIC, 2017). Furthermore, it aids in the safeguarding of plant leaves against the harmful effects of ultraviolet

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radiation B (UV-B) (GOULD et al., 2000; ESPÍNDOLA et al., 2019; AYGUL et a., 2022).

Extensive investigations, encompassing both in vitro and in vivo experiments, have substantiated the multifarious physiological effects of CAA and its derivatives. These effects include but are not limited to antibacterial activity (GENARO-MATTOS et al., 2015), anti-inflammatory and antiviral activity (RODRIGUES et al., 2015), cardioprotective and antioxidant activity (AGUNLOYE et al., 2019), antiproliferative and anti-atherosclerotic activity (NAGAOKA et al., 2002; VERMA and HANSCH, 2004), hepatoprotective activity (YANG et al., 2013), anticancer activity (LEE et al.i 2008; MCGLYNN et al., 2015), and anti-hepatocellular carcinoma activity (LEE et al.i 2008; WON et al., 2010). Of particular significance among these properties is the highlighted anti-hepatocarcinoma activity, given the status of hepatocarcinoma (HCC) as a leading cause of cancerrelated mortality worldwide (MCGLYNN et al., 2015). Consequently, further exploration concerning the chemical and pharmacological aspects of CAA is imperative to pave the way for potential new drug development and, subsequently, expand therapeutic avenues (ZHANG et al., 2017).

Carbonic anhydrases (CA, EC 4.2.1.1.) play a pivotal role in facilitating the conversion between carbon dioxide (CO₂) and bicarbonate (HCO₃⁻) and the subsequent dehydration of bicarbonate, resulting in the regeneration of CO₂ within an acidic milieu (SUPURAN, 2008; FIDAN et al., 2015; YAKAN et al., 2023).

In mammals, a total of sixteen CA isozymes have been identified thus far, with notable emphasis on CA II and CA IX as highly efficient catalysts for carbon dioxide hydration (SUPURAN, 2008; SUPURAN, 2017; ARSLAN et al., 2016; ARSLAN et al., 2020). CA I and II is primarily found in erythrocytes, but also exhibits distribution in numerous secretory tissues of the gastrointestinal tract, kidneys, lungs, eyes, central nervous system, and more. Conversely, CA IX represents a tumor-associated isoform (SUPURAN, 2008; ABDEL-AZIZ et al., 2015; DIZDAROGLU et al., 2020; YAKAN et al., 2023). Moreover, several other CA isoforms have been identified in diverse tissues, actively participating in vital biological processes including acid-base homeostasis, respiration, carbon dioxide and ion transport, bone resorption, ureagenesis, gluconeogenesis, lipogenesis, and electrolyte secretion. The multitude of CA isozymes involved in these intricate processes hold significant therapeutic potential as they emerge as promising targets for modulation, whether through inhibition or activation, in the treatment of various disorders such as edema, glaucoma, obesity, cancer, and epilepsy (SUPURAN, 2008; SUPURAN, 2017; URCAR et al., 2016; YAKAN et al., 2023).

Glutathione reductase (GR), an enzyme categorized as EC 1.6.4.2, plays a crucial role in maintaining the equilibrium of the intracellular redox system. Its primary function involves facilitating the conversion oxidized glutathione (GSSG) to reduced of glutathione (GSH) through the utilization of nicotinamide adenine dinucleotide phosphate (NADPH) molecules (KARPLUS et al., 1989; COBAN et al., 2007; URCAR et al., 2016). Extensive research has been conducted to explore the inhibition of the GR enzyme using various compounds. The findings of these investigations have consistently indicated a reduction in the GSH/GSSG ratio and an elevation in NAD(P)H/NAD(P)⁺ the ratio (SENTURK et al., 2008; COUTO et al., 2016; KOCAOGLU et al., 2019; USTUNDAG et al., 2022). Although some studies have demonstrated that GR inhibition does not influence the generation of free radicals or the expression of other enzymes involved in GSH biosynthesis, it is worth noting that the potent activity of the GR enzyme in cancer cells contributes to their resistance against several chemotherapeutic drugs. Increased intracellular GSH levels offer a significant survival advantage for tumor cells, particularly in lung, breast, larynx, colon, and bone marrow cancers. Consequently, the inhibition of the GR enzyme presents a promising avenue for anticancer treatment, considering the potential of such substances (BALENDIRAN et al., 2004; SENTURK TRAVERSO et et al.. 2009: al.. 2013: SANTACROCE et al., 2023).

In this study, the potential of CAA to inhibit human CA I, II, IX and GR enzymes was tried to be determined. For this purpose, both theoretical (embedding) and experimental experiments were carried out. It also aimed to explore the potential therapeutic effects of CAA on gastric cancer cells and keratinocyte normal cells.

2. Materials and Methods

Chemicals

hCA I (C4396), hCA II (C6165), and hCA IX (CA 9, human recombinant (SRP6483) were procured from Sigma-Aldrich company. All other chemicals and solvents were purchased from Merck (Darmstadt, Germany).

Measurement of Glutathione Reductase Activity

To assess the enzymatic activity of GR, a spectrophotometric method is employed, which involves monitoring the decrease in NADPH levels when the substrate GSSG is present. This reduction in NADPH concentration is measured specifically at a wavelength of 340 nm (BEUTLER, 1984; USTUNDAG et al., 2022).

Measurement of Carbonic Anhydarase Activity

The inhibitory activity of CAA on hCA I, hCA II, and hCA IX were determined according to the esterase method (VERPOORTE et al., 1967; ARSLAN et al. 2020). In inhibitory studies, p-nitrophenyl acetate was employed as the substrate.

Inhibitory Effect Determination of IC₅₀ Values of CAA

With inhibitor studies, the activity of CAA, whose solutions were prepared, were added to the cuvette in different concentrations and their activities were measured. CAA were plotted as % Activity- [I], IC₅₀ values were calculated from the equation of the curve.

In silico Studies

In silico placement studies were conducted to investigate interactions between CAA and amino acid residues around the active site of the CA I, II, and IX enzyme. All pdb files were obtained from rcsb.org. The AutoDockTools1 program (version 1.5.7) was used for the preparation of all ligands and enzymes. Autodock Vina2 program (version 1.1.2) was used for all docking experiments, the entire surface of each enzyme was investigated, the exhaustiveness value was set as 32, the energy_range value was assigned as 5, and the best 5 results were asked to be listed. Twelve trials of each molecule were made for each

enzyme, and the highest scoring conformation of these was aligned with the receptor protein in the PyMOL-oss3 program (version 2.4.1) (SANNER, 1999; TROTT and OLSON, 2010). The evaluation of the interactions was carried out in the Biovia Discovery Studio Visualizer4 (version 21.1.0.2029) program. Docking scores are summarized in Table 1, Figure 2, 3, and 4.

AGS and HaCaT Cell Line Culture Studies

AGS gastric cancer and HaCaT keratinocyte normal cell line attached to the base of the cell culture layer were passaged when their cell density reached 70-80%. Cells were washed with 10 ml of PBS and treated with 4 ml of 0.25% Trypsin/EDTA solution (Gibco) for 2-3 minutes in the incubator to lift them off the flask base and the separation of cells was also observed under the microscope. Trypsin activity was stopped by adding 2 ml of FBS. The cells were then transferred to a 15 ml centrifuge tube and centrifuged at 1800 rpm for 10 minutes at room temperature. The supernatant was completely discarded and the cell pellet was dissolved in 10 ml of DMEM-LG. The mixture was then centrifuged again at 1800 rpm for 10 minutes at room temperature. The supernatant was discarded again, and the cell pellet was dissolved in the appropriate volume of DMEM-LG, and the cells were stained with Trypan blue and counted with a hemocytometer. Depending on the number of cells, cells were seeded into new cell culture dishes in DMEM-LG containing 10% FBS and 0.1 ml/ml Primocin, and the dishes were placed in an incubator at 37°C containing 5% CO₂. The MCF-7 cell line was passaged again when they reached 70-80% density (STRIEDINGER et al., 2021).

Cell Viability Test (MTS)

The passaged cells were counted and seeded in two separate 96-well plates with 1x104 cells in each well. After 24 hours of incubation, CAA was given to the cells in the wells at certain concentrations (500, 250, 100, 50, 25 μ M). Each concentration was run in 3 replicates. Cell viability test (MTS) was performed after 24 hours. Absorption was measured in the ELISA reader at 450 nm. The graph was drawn according to the resulting absorption values (KOCANCI et al., 2017).

3. Results and Discussion

GR, an antioxidant enzyme, carries out a vital function by regulating the redox metabolism of GSH within the cellular framework of numerous organisms. Its involvement in the modulation of GSH redox hemostasis contributes to the synthesis of deoxyribonucleotides. Moreover, peroxide plays a critical role in the detoxification of 2-oxoaldehydes and xenobiotics. The viability of rapidly dividing cells and those resilient to oxidative stress is significantly reliant on the replenishment of GSH. Consequently, the identification of potential GR inhibitors assumes paramount importance in the of antitumor advancement and antiparasitic pharmaceutical agents (KARPLUS et al., 1989; BOEHME et al., 2000; USTUNDAG et al., 2022).

In investigations concerning the human erythrocyte GR enzyme, which serves as a model for drug trials, it has been ascertained that nitro aromatic compounds containing a quinoline moiety, commonly employed as antisitulant and anticancer medications, exert an influence on the activity of GR enzyme (GRELLIER et al., 2001). Furthermore, diverse substances encompassing certain drugs, metal ions, and nitro groups have been discovered to possess inhibitory properties against GR enzymes derived from various sources (GRELLIER et al., 2001; COBAN et al., 2007; CAKMAK et al., 2011; KOCAOGLU et al., 2019).

In trials for human CA I, II, IX and GR enzymes with CAA. The IC₅₀ value for hCA I was determined as 10.26 μ M, for hCA II as 9.14 μ M, for hCA IX as 8.96 μ M and for GR as 25.84 μ M. Acetazolamide (AZA) was used as the reference for the tested CA enzymes, and N,N-bis(2-chloroethyl)-N-nitrosourea (BCNU) was used for the GR. AZA was observed as 36.2 μ M for hCA I, 0.37 μ M for hCA II and 0.93 micromolar for hCA IX. For GR it was observed as 465 μ M for BCNU.

The utilization of CA inhibitors in the management of several conditions, such as cancer, glaucoma, and obesity, is well-established. Consequently, the quest for innovative and potent molecular frameworks for the therapeutic intervention of these ailments assumes a noteworthy strategy (ISIK et al., 2015; SUPURAN, 2017; SENTURK et al., 2022).

Three methods were used to determine the solubility of the molecule in water in Swiss-ADME. These are the ESOL, Ali and SILICOS-IT methods. Caffeic acid was estimated to be water soluble according to three methods. In pharmacokinetic estimates, gastrointestinal (GI) absorption was predicted to be high. In the drug similarity part, it shows that the CAA may be active in four of the analyzes of proprietary chemical collections from five different major pharmaceutical companies. According to the estimates of Medicinal Chemistry; It should be used with caution in human therapy as it gives a warning for PAINS (pan assay interference compounds). Bcolor analysis (to be toxic by default, chemically

reactive, metabolically unstable or having properties responsible for poor pharmacokinetics) has given two warnings. In this case, direct use in the body should be well adjusted. There is no similar molecule. Systemic acceptability score is good.

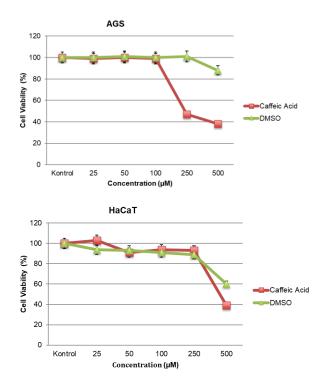


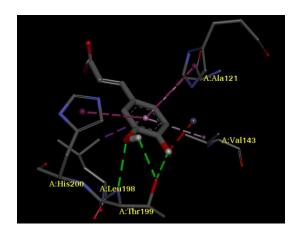
Figure 1. Cell viability result graph of AGS and HaCaT cells treated with 500, 250, 100, 50, 25 μ M caffeic acid.

The passaged cells were counted and seeded in two separate 96-well plates with 1x104 cells in each well. After 24 hours of incubation, CAA was given to the cells in the wells at certain concentrations (500, 250, 100, 50, 25 μ M). Each concentration was run in 3 replicates. Cell viability test (MTS) was performed after 24 hours. Absorption was measured in the ELISA reader at 450 nm. According to the result in Figure 1, in gastric cancer (AGS) cell line, CAA reduced cell viability to 47% and 38%, respectively, at 250 and 500 μ M concentrations. As the dose increased, the toxic effect on the gastric cancer cell line also increased. These substances were also tested on a normal human keratinocyte cell line (HaCaT). According to the graph, CAA showed toxic effect only at the highest dose (500 μ M), reducing cell viability to 39%.

Using SwissADME, we attempted to calculate physicochemical descriptors as well as predict ADME parameters, pharmacokinetic properties, drug-like nature and medicinal chemistry friendliness of one or more small molecules to support drug discovery (Figure 5).

Table 1. Docking scores and K_i values of caffeic acid for tested enzymes.

Enzyme (PDB	Affinity	K _i value
ID)	(kcal/mol)	(µM)
hCA I (2CAB)	-6,2	5.12
hCA II (3KS3)	-6,5	4.86
hCA IX (6FE2)	-7,2	4.25



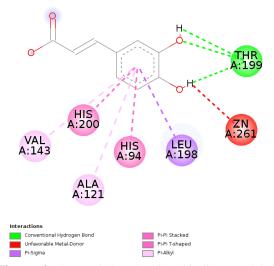


Figure 2. 3D and 2D docking binding models of caffeic acid with hCA I enzyme, respectively.

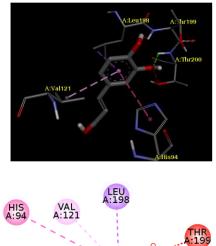




Figure 3. 3D and 2D docking binding models of caffeic acid with hCA II enzyme, respectively.

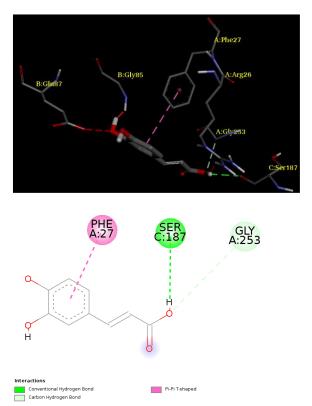
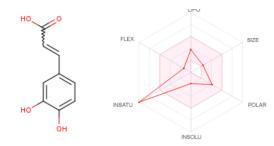


Figure 4. 3D and 2D docking binding models of caffeic acid with hCA IX enzyme, respectively.



SMILES OC(=O)/C=C/c1ccc		
	sicochemical Properties	
Formula	C9H8O4	
Molecular weight Num. heavy atoms	180.16 g/mol 13	
Num. arom. heavy atoms	6	
Fraction Csp3	0.00	
Num. rotatable bonds	2	
Num. H-bond acceptors	4	
Num. H-bond donors	3	
Volar Refractivity	47.16	
TPSA 📀	77.76 Ų	
	Lipophilicity	
.og P _{o/w} (iLOGP) 📀	0.97	
.og P _{o/w} (XLOGP3) 😣	1.15	
-		
og P _{o/w} (WLOGP)	1.09	
₋og P _{o/w} (MLOGP) [©]	0.70	
₋og P _{o/w} (SILICOS-IT) [©]	0.75	
Consensus Log P _{o/w} 🥺	0.93	
	Water Solubility	
Log S (ESOL) 📀	-1.89	
Solubility	2.32e+00 mg/ml ; 1.29e-02 mol/l	
Class 📀	Very soluble	
Log S (Ali) 📀	-2.38	
Solubility	7.55e-01 mg/ml ; 4.19e-03 mol/l	
Class 📀	Soluble	
Log S (SILICOS-IT) 📀	-0.71	
Solubility	3.51e+01 mg/ml ; 1.95e-01 mol/l	
Class 📀	Soluble Pharmacokinetics	
GI absorption 🤨	High	
BBB permeant 0	No	
P-gp substrate 0		
CYP1A2 inhibitor 📀	No	
CYP2C19 inhibitor 🥹	No	
CYP2C9 inhibitor 🔮		
CYP2D6 inhibitor 🥹	No	
CYP3A4 inhibitor 😔	No	
Log K _p (skin permeation) 🤨	-6.58 cm/s	
ininaki 🙆	Druglikeness	
Lipinski 📀	Yes; 0 violation	
Ghose 🤨 Veber 😗	Yes	
10001	100	
Egan 😟	Yes	
Muegge 🤨	No; 1 violation: MW<200	
Bioavailability Score 🧐	0.56	
DAING 0	Medicinal Chemistry	
PAINS ()	1 alert: catechol_A 🥹	
Brenk 📀	2 alerts: catechol, michael_acceptor_1	
Leadlikeness 📀	No; 1 violation: MW<250	
Synthetic accessibility 📀	1.81	

Figure 5. SwissADME result for caffeic acid.

4. Conclusion

In this investigation, the primary objective is to employ a dual methodology involving in silico and in vitro analyses. The *in silico* assessment will provide valuable insights into the potential interactions of CAA with specific metabolic enzymes. Through this computational approach, the molecular interactions and binding affinities between caffeic acid and target enzymes will be evaluated, offering a preliminary understanding of the possible effects.

Overall, this study seeks to shed light on the in silico and *in vitro* effects of CAA on metabolic enzymes, while simultaneously exploring its potential therapeutic implications in AGS gastric cancer cells and HaCaT keratinocyte normal cells. Through the integration of computational and experimental approaches, this investigation strives to contribute to our understanding of the molecular mechanisms underlying CAA's multifaceted properties and its potential application in cancer therapy.

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References

- ABDEL-AZIZ AA-M, EL-AZAB AS, EKINCI D, SENTURK M, SUPURAN CT. 2015. Investigation of arenesulfonyl-2imidazolidinones as potent carbonic anhydrase inhibitors. J Enzym Inhib Med Chem. 30(1):81-84.
- AGUNLOYE OM, OBOH G, ADEMILUYI AO, ADEMOSUN AO, AKINDAHUNSI AA, OYAGBEMI AA, et al. 2019. Cardioprotective and antioxidant properties of chlorogenic caffeic acid and acid: mechanistic role of angiotensin converting cholinesterase and arginase enzyme, activities in cyclosporine induced hypertensive rats. Biomed Pharmacother. 109:450-458.

- ARSLAN T, CELIK G, CELIK H, SENTURK M, YAYLI N, EKINCI D. 2016. Synthesis and Biological Evalution of Novel Bischalcone Derivatives as Carbonic Anhydrase Inhibitors. Archive der Pharmazie. 349(9):741-748.
- ARSLAN T, TURKOGLU EA, SENTURK M, SUPURAN CT. 2016. Novel chalcone substituted benzenesulfonamides as inhibitors for human carbonic anhydrases. Bioorg Med Chem Lett. 26(24):5867-5870.
- ARSLAN T, CEYLAN MB, BAS H, BIYIKLIOGLU Z, SENTURK M. 2020. Design, Synthesis, Characterization of Novel Peripherally Tetra 1,2,3-Triazole Substituted Phthalocyanines and Inhibitor Effect on Cholinesterases (AChE/BChE) and Carbonic Anhydrases (hCA I, II and IX). Dalton Trans. 49:203-209.
- AYGUN B, SENTURK M, CINAN E, SIMSEK O, SAYYED MI, KARABULUT A. 2022. Determination of Biological Radioprotective Characteristics of some natural organic compounds for Radiation Shielding Applications. Radiochimica Acta. https://doi.org/10.1515/ract-2022-0028.
- BALENDIRAN GK, DABUR R, FRASER D. 2004. The role of glutathione in cancer. Cell Biochem Funct 22:343-352.
- BEUTLER E. 1984. Red cell metabolism. A manual of biochemical methods. Grune and Stratton Inc, Orlando. ISBN: 0808916726 9780808916727.
- BOEHME CC, ARSCOTT LD, BECKERK, SCHIRMER RH, WILLIAMS CH. 2000. Kinetic characterization of glutathione reductase from the malarial parasite Plasmodium falciparum Comparison with the human enzyme. J Biol Chem. 275:37317-37323.
- CAKMAK R, DURDAGI S, EKINCI D, SENTURK M, TOPAL G. 2011. Design, synthesis and biological evaluation of novel nitroaromatic compounds as potent glutathione reductase inhibitors. Bioorg Med Chem Lett. 21;5398-5402.
- CHEN JH, HO CT. 1997. Antioxidant activities of caffeic acid and its related hydroxycinnamic

acid compounds. J Agric Food Chem. 45:2374-2378.

- COBAN TA, SENTURK M, CIFTCI M, KUFREVIOGLU OI. 2007. Effects of some metal ions on human erythrocyte glutathione reductase: an in vitro study. Protein Peptide Lett. 14(10)1027-1030.
- COUTO N, WOOD J, BARBER J. 2016. The role of glutathione reductase and related enzymes on cellular redox homoeostasis network. Free Rad Biol Med. 95:27-42.
- DEMIRDAG R, COMAKLI V, SENTURK M, EKINCI D, KUFREVIOGLU OI, SUPURAN CT. 2013. Purification and characterization of carbonic anhydrase from sheep kidney and effects of sulfonamides on enzyme activity. Bioorg Med Chem. 21(6):1522-1525.
- DIZDAROGLU Y, ALBAY C, ARSLAN T, ECE A, TURKOGLU EA, EFE A, SENTURK M, SUPURAN CT, EKINCI D. 2020. Design, synthesis and molecular modelling studies of some pyrazole derivatives as carbonic anhydrase inhibitors. J Enzym Inhib Med Chem. 35(1):289-297.
- ESPÍNDOLA KMM, FERREIRA RG, NARVAEZ LEM, SILVA ROSARIO ACR, DA SILVA AHM, SILVA AGB, VIEIRA AP, MONTEIRO MC. 2019. Chemical and Pharmacological Aspects of Caffeic Acid and Its Activity in Hepatocarcinoma. Front. Oncol. 9:541.
- FIDAN I, SALMAS RE, ARSLAN M, SENTURK M, DURDAGI S, EKINCI D, SENTURK E, COSGUN S. SUPURAN CT. 2015. Carbonic anhydrase inhibitors: design, synthesis, kinetic, docking and molecular dynamics analysis of novel glycine and phenylalanine sulphonamide derivatives. Bioorg Med Chem. 23(23):7353-7358.
- GENARO-MATTOS TC, MAURÍCIO ÂQ, RETTORI D, ALONSO A, HERMES-LIMA M. 2015. Antioxidant activity of caffeic acid against iron-induced free radical generation-A chemical approach. PLoS ONE. 10:e0129963.
- GOULD KS, MARKHAM KR, SMITH RH, GORIS JJ. 2000. Functional role of anthocyanins in

the leaves of Quintinia serrata A. Cunn. J Exp Bot. 51:1107-1115.

- GRELLIER P, SARLAUSKAS J, ANUSEVICIUS Z, MAROZIENE A, HOUEE-LEVIN C. SCHREVEL J. CENAS N. 2001. Antiplasmodial activity of nitroaromatic and quinoidal compounds: redox potential vs. inhibition of erythrocyte glutathione reductase. Arch Biochem Biophys. 393:199-206.
- ISIK S, VULLO D, DURDAGI S, EKINCI D, SENTURK M, CETIN A, SENTURK E, SUPURAN CT. 2015. Interaction of carbonic anhydrase isozymes I, II, and IX with some pyridine and phenol hydrazinecarbothioamide derivatives. Bioorg Med Chem Lett. 25(23):5636-5641.
- KARPLUS PA, PAI EF, SCHULZ GE. 1989. A crystallographic study of the glutathione binding site of glutathione reductase at 0.3 nm resolution. Eur J Biochem. 178:693-703.
- KOCANCI FG, HAMAMCIOGLU B, ASLIM B. 2017. The anti-AChE and antiproliferative Activities of Glaucium acutidentatum and Glaucium corniculatum Alkaloid Extracts. J App Pharm Sci. 7(8):191-200.
- KOCAOGLU E, TALAZ O, CAVDAR H, SENTURK M, SUPURAN CT, EKINCI D. 2019. Determination of the inhibitory effects of N-methylpyrrole derivatives on glutathione reductase enzyme. J Enzym Inhib Med Chem. 34:51-54.
- LEE KW, KANG NJ, KIM JH, LEE KM, LEE DE, HUR HJ, et al. 2008. Caffeic acid phenethyl ester inhibits invasion and expression of matrix metalloproteinase in SK-Hep1 human hepatocellular carcinoma cells by targeting nuclear factor kappa B. Genes Nutr. 2:319-322.
- MCGLYNN KA, PETRICK JL, LONDON WT. 2015. Global epidemiology of hepatocellular carcinoma: an emphasis on demographic and regional variability. Clin Liver Dis. 19(2):223-238.
- NAGAOKA T, BANSKOTA AH, TEZUKA Y, SAIKI I, KADOTA S. 2002. Selective antiproliferative activity of caffeic acid phenethyl ester analogues on highly liver-Metastatic murine colon 26-L5 carcinoma

cell line. Bioorganic Med Chem. 10:3351-3359.

- RODRIGUES JL, ARAÚJO RG, PRATHER KLJ, KLUSKENS LD, RODRIGUES LR. 2015. Heterologous production of caffeic acid from tyrosine in Escherichia coli. Enzyme Microb Technol. 71:36-44.
- SANNER MF. 1999. Python: a programming language for software integration and development. J Mol Graph Model. 17(1):57-61.
- SANTACROCE G, GENTILE A, SORIANO S, NOVELLI A, LENTI MV, DI SABATINO A. 2023. Glutathione: Pharmacological aspects and implications for clinical use in non-alcoholic fatty liver disease. Front Med (Lausanne). 22;10:1124275.
- SENTURK E, GELEN V, SENGUL E, SENTURK M, YILDIRIM S, TEKIN S. 2022. Determination of the Protective Effects of Hesperidin and Curcumin on the Inhibition Effect of 5-Fluorouracil in Mouse Kidney Carbonic Anhydrase Enzyme. Acta Physiol. 234:68.
- SENTURK M, KUFREVIOGLU OI, CIFTCI M. 2008. Effects of some antibiotics on human erythrocyte glutathione reductase: An in vitro study, J Enzym Inhib Med Chem. 23(1):144-148.
- SENTURK M, KUFREVIOGLU OI, CIFTCI M. 2009. Effects of some analgesic and anestethic drugs on human erythrocyte glutathione reductase: an in vitro study. J Enzym Inhib Med Chem. 24(2):420-424.
- SILVA T, OLIVEIRA C, BORGES F. 2014. Caffeic acid derivatives, analogs and applications: a patent review (2009-2013). Expert Opin Ther Pat. 24:1257-1270.
- STRIEDINGER K, BARRUET E, POMERANTZ JH. 2021. Purification and preservation of satellite cells from human skeletal muscle. STAR Protoc. 29;2(1):100302.
- SUPURAN CT. 2008. Carbonic anhydrases: novel therapeutic applications for inhibitors and activators. Nature Rev Drug Discov. 7:168-181.
- SUPURAN CT. 2017. Carbonic anhydrase inhibition and the management of hypotoxic tumours. Metabolites. 7:48-61.

- SwissADME: a free web tool to evaluate pharmacokinetics, drug-likeness and medicinal chemistry friendliness of small molecules. Sci Rep. 2017:7:42717.
- TOSOVIC J. 2017. Spectroscopic features of caffeic acid: theoretical study. Kragujev J Sci. 3918025435:99-108.
- TRAVERSO N, RICCIARELLI R, NITTI M, MARENGO B, FURFARO AL, PRONZATO MA, MARINARI UM, DOMENICOTTI C. 2013. Role of glutathione in cancer progression and chemoresistance. Oxid Med Cell Longev. 2013:972913.
- TROTT O, OLSON AJ. 2010. AutoDock Vina: improving the speed and accuracy of docking with a new scoring function, efficient optimization, and multithreading. J Comp chem. 31(2):455-461.
- URCAR H, SENTURK E, SENTURK M, GUL M, YILDIRIM S. 2016. Investigation of Effects of Some Catecholamines on The Activity of Carbonic Anhydrase Enzyme Purified From Bovine Kidney Tissue. Acta Physiol. 218:57.
- USTUNDAG H, SENTURK E, SENTURK M. 2022. Effect of White Noise and Light Exposure on Rat Testis Glutathione Reductase Enzyme. Acta Physiol. 234:79.
- VERMA RP, HANSCH C. 2004. An approach towards the quantitative structure-activity relationships of caffeic acid and its derivatives. ChemBioChem. 5:1188-1195.
- VERPOORTE JA, MEHTA S, EDSALL JT. 1967. Esterase activities of human carbonic anhydrases B and C. J Biol Chem. 242(18):4221-4229.
- VULLO D., INNOCENTI A., NISHIMORI I., PASTOREK J., SCOZZAFAVA A., PASTOREKOVA S., SUPURAN CT. 2005. Carbonic anhydrase inhibitors. Inhibition of the transmembrane isozyme XII with sulfonamides-a new target for the design of antitumor and antiglaucoma drugs? Bioorg. Med. Chem. Lett. 15:963-969.
- WON C, LEE CS, LEE JK, KIM TJ, LEE KH, YANG YM, et al. 2010. CADPE suppresses cyclin D1 expression in hepatocellular

carcinoma by blocking IL-6-induced STAT3 activation. Anticancer Res. 30:481-488.

- YAKAN H, BILIR G, ÇAKMAK Ş, TAŞ O, TÜRKÖZ KARAKULLUKÇU N, SOYDAN E, KUTUK H, GUCLU C, SENTURK M, ARSLAN T, OZTURK S, AKSAKAL E, EKINCI D. 2023. Inhibitory effects of sulfenimides on human and bovine carbonic anhydrase enzymes. J Enzyme Inhib Med Chem. 38(1):2194573.
- YANG SY, HONG CO, LEE GP, KIM CT, LEE KW. 2013. The hepatoprotection of caffeic acid and rosmarinic acid, major compounds of Perilla frutescens, against t-BHP-induced oxidative liver damage. Food Chem Toxicol. 55:92-99.
- ZHANG Z, WANG D, QIAO S, WU X, CAO S, WANG L, ET al. 2017. Metabolic and microbial signatures in rat hepatocellular carcinoma treated with caffeic acid and chlorogenic acid. Sci Rep. 7:4508.