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REVIEW ARTICLE DERLEME MAKALESİ

# HIGH HYDROSTATIC PRESSURE TREATMENT OF FRUIT, FRUIT PRODUCTS AND FRUIT JUICES: A REVIEW ON PHENOLIC COMPOUNDS

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#### Abstract:

Phenolic compounds are healthy substances, therefore amount of phenolic compounds in fruits/fruit juices are important for customers. Flavonoids are the most common and widely distributed group of plant phenolics. Fruits and fruit juices are among the best sources of polyphenols in the human diet because of their high content in most fruits. Processing fruits can change their content of phenolic compounds and heat treatments usually decrease phenolic compounds in fruits. Minimising losses of these bioactive compounds during processing and storage are one of the priorities and challenge for manufacturers. High hydrostatic pressure (HHP) is a technology which can be used instead of thermal process such as pasteurization and can be combined with thermal process for sterilization. This study discusses the impact of HHP processing

conditions on the stability of phenolic compounds in fruits and their products after process, during storage and compares with thermal process. Matrix of food and processing parameters such as pressure, temperature, time are important for retaining of phenolic compounds. In general, several authors have proven that HHP treatment on different fruit products slightly modified their content of phenolic compounds. There has not been much research comparing HHP with thermal processes on effects of phenolic compounds in fruits and fruit products, and researches generally indicate that HHP is better than thermal processes on retaining phenolic compounds.

**Keywords:** Polyphenol, High hydrostatic pressure, Fruit, Anthocyanin, Thermal treatment

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### Introduction

Consumers are interested in the nutritional and health-related characteristics of fruits and vegetables. Phenolic compounds are healthy substances, so amount of phenolic compounds in fruit and vegetable products is important for customers. Over the past decade much research has been carried out on plant phytochemicals and their positive health effects. It has been demonstrated that a diet rich in fruits and vegetables containing various classes of polyphenols (phenolic acids, flavonols, catechin monomers, proanthocyanidins, flavones, flavanones, and anthocyanins) decreases the risk of premature mortality, cardiovascular disorders, age-induced advancing oxidative inflammatory responses and diverse degenerative diseases (Michels et al., 2000; Torres et al. 2011; Sánchez-Moreno et al. 2009).

Flavonoids are the most common and widely distributed group of plant phenolics. Among them, flavones, flavonols, flavanols, flavanones, anthocyanins, and isoflavones are particularly common in fruits. Fruits are among the best sources of polyphenols in the human diet because of their high content in most fruits and the relatively large serving sizes (100-200 g).

High hydrostatic pressure (HHP) is a technology which subject's liquid and solid foods, with or without packaging, to pressures between 100 and 800 MPa. Process temperature during pressure treatment can be specified from below 0°C (to minimize any effects of adiabatic heat) to above 100°C. High pressure processing (HPP), is also described as high hydrostatic pressure (HHP), or ultra high pressure (UHP) processing in literature (FDA, 2015).

HHP technology can be used for various purposes such as pasteurization, extraction and osmotic dehydration. HHP technology can be combined with other thermal and non-thermal technologies such as ultrasound, pulsed electrical field (PEF), gamma-irradiation etc., which are excluded in this paper. This paper aims to provide a detailed and critical review of the latest applications of HHP on fruit and fruit juices, the effect of HHP on phenolic compounds are especially emphasized.

### **Thermal Processing and HHP**

Thermal processing, which is the conventional method for the inactivation of plant enzymes and inhibition of microorganisms, is not suitable for soft fruits. In addition to desirable effect of microbial and enzyme inactivation, it results in the loss of physicochemical and nutritional quality attributes of the product. The color and flavor of most fruits deteriorate during thermal processing. As a consequence, the addition of artificial color is usually necessary during thermal processing of fruits (Burrows, 2001).

The application of HHP has the potential to produce high quality fruits that display characteristics of fresh foods, microbiologically safe and extended shelf life (Huang et al., 2013). HHP preserves nutritional value with a minimal effect on the product quality and sensory properties of fruits and vegetables, since low molecular weight food compounds, such as flavoring agents, pigments, and some vitamins are not altered, because covalent bonds are not affected by pressure (Oey et al., 2008; Bala et al., 2008; Ramirez at al., 2009). Over the last years, has been investigated and several commercial products, including fruit juices, i.e mandarin, grapefruit, apple, orange, carrot juices and broccoli-apple juice mixture treated by HHP are currently available on market (Ahmed et al., 2005; Barba et al., 2011; Bull et al., 2004; McInerney et al., 2007).

The effect of HHP can vary depending on processing conditions (pressure, hold time, temperature), food form (whole, pieces, juice, puree, mousse or smoothie) and intrinsic factor of food such as pH. Grape, strawberry, apricot and apple polyphenol oxidase (PPO) seem to be more pressure sensitive than other PPO's. Apricot, strawberry and grape PPO can be inactivated by pressures exceeding 100, 400 and 600 MPa, respectively. Depending on pH, pressures of 100-700 MPa were needed for the inactivation of apple PPO. There are many studies on thermal denaturation of PPO, but less data about the pressure inactivation of PPO are available. It is known that PPO may be activated or inactivated by HHP (Lopes et al., 2010).

HHP-processed fruits and fruit products may exhibit long-term changes in flavor, nutritional properties, and color as a result of the residual activity of endogenous enzymes (Boynton et al., 2002). To reduce enzyme activity and preserve food quality during processing and storage, different methods have been successfully applied in combination with HHP, e.g. low pH conditions, refrigerated storage, high temperature treatment,

and anti-browning agents (Guerrero-Beltrán et al., 2006; Krebbers et al., 2003).

## **Effects of HHP on Phenolics of Fruits and Fruit Products**

Processing fruits and fruit products can change their content of phenolic compounds and heat treatments can reduce phenolic compounds in fruits and fruit products. Many factors affect the stability of the anthocyanins, including temperature, pH, oxygen, enzymes, the presence of copigments and metallic ions, ascorbic acid, sulphur dioxide as well as sugars and their degradation products. Similarly, the behaviour of the polyphenol content is similar to that observed for the anthocyanin content in several liquids, semi-solid or solid foodstuffs (Ferrari et al., 2011). Great technological and research efforts have been made to obtain fruit juices by HHP without the quality and nutritional damage caused by heat treatments. Foods are processed in batch systems, in a pressure range of 50-1000 MPa; process temperature during pressure treatment can be from below 0 to above 100 °C, while exposure time usually ranges from seconds to 20 minute (Corbo et al., 2009).

HHP treatment at ambient temperature is reported to have minimal effect on the anthocyanins content of various fruits and vegetables. Several authors have reported that the anthocyanins of different liquid foods are stable to HHP treatment at moderate temperatures. Impact of processing conditions affects the stability of phenolic compounds in fruits and their products. Matrix of food and processing parameters are important for retaining of phenolic compounds. Combination with other mild processing methods (ultrasound, gamma-irradiation, antimicrobial peptides) can help to retain nutritional and health-related characteristics of plant-derived products by assuring the safety of the product indeed during the storage period. The optimization of the HHP process conditions (pressure, temperature and time) is important for phenolic quantification (Tokusoglu et al., 2010; Corbo et al., 2009).

Several authors have reported the increased extractability of coloured pigments in food components at extreme pressures results polyphenol content increase. The authors attribute the increase of the total phenolic content to the release of some antioxidant components such as anthocyanins, amino acids and protein with phenolic hydroxyl group after HHP treatment

from solid suspended particles following the high pressure processing. For semi-solid foods, the HHP treatment could potentially improve the total polyphenol content and prevent the degradation deriving from the application of temperatures, as observed in traditional thermal treatments for food pasteurization/sterilization. This increase in total phenolic content (TPC) may be related to an increased extractability of some of the antioxidant components following high pressure processing. According to Le Chatelier's theory, the volume of system tends to be reduced during the pressure promoting period. In this process, the extracting solvent comes into cells to integrate with bioactive components. Besides, HHP can increase the rate of mass transfer resulting in an enhancement of solvent penetration into the cells by disrupting the cellular walls and hydrophobic bonds in the cell membrane, which may lead to a high permeability (Jun et al., 2009; Prasad et al., 2009; Cao et al., 2011).

Most of the studies showed that total phenolic compounds and some phenolic compounds (hesperetin, naringenin and anthocyanins) increased or unchanged by HHP. In general, several authors have proven that HHP on different fruit products (orange juice, lemon juice, apple strawberry puree, kiwifruit orange/carrot juice, apple/broccoli juice) slightly modified their content of phenolic compounds (Sánchez-Moreno et al., 2009; Cao et al., 2011; Barba et al., 2012). However, the effect of high pressure on the anthocyanin content of fruit derivatives cannot be generalized, while the composition of the product, the activity of the oxidative enzymes and the processing and operative conditions could compromise the efficiency of the HHP treatment. Enzymes such as polyphenoloxidase, peroxidase and β-glucosidase have been implicated in the degradation of phenolics and anthocyanins (Torres et al., 2011). oxidase (PPO) Polyphenol catalyses hydroxylation of monophenols to o-diphenols and the oxidation of o-diphenols to o-quinones in the presence of molecular oxygen. The o-quinones are highly unstable and either react with high molecular weight polymers or form macromolecular complexes with amino acids and proteins (Terefe et al., 2009; Thomas-Barberan et al., 2001). Peroxidase (POD) catalyses the oxidation of phenolic compounds in the presence of hydrogen peroxides. Since the internal concentration of hydrogen peroxide in plants is low, the role of peroxidase in the in vivo degradation of

polyphenols is sometimes questioned. However, a synergistic activity of PPO and POD is possible with PPO acting as the promoter of POD since hydrogen peroxide is generated during the PPO catalysed oxidation of phenolic compounds (Thomas-Barberan et al., 2001).

### Effects of HHP on Anthocyanin Content of Fruit and Fruit products

Anthocyanins were usually reported to be unstable, especially at high and abuse temperatures during processing and storage. Changes of phenolics and anthocyanins in fruits after HHP process are showed in Table 1. No significant change of anthocyanins is observed in strawberry halves after 600 MPa HHP process. Changes of phenolics and anthocyanins in fruits juices, nectars and smoothies after HHP process are showed in Table 2. HHP effectively retained anthocyanins, phenolic compounds and color of pomegranete juice for treated samples with 350 MPa and 550 MPa at room temperature (Varela-Santos et al., 2012). Alpas (2013) also observed no significant decrease in monomeric anthocyanin pigment concantrations for HHP samples, but thermal treatment (85°C/10min.) decreased significantly monomeric anthocyanin pigment concantrations in pomegranate juices. Greater retention of HHP samples is found compared to HTST  $(110^{\circ}\text{C/8.6s})$ samples cloudy

pomegranate juice (Chen et al., 2013). In another study, 63% of pomegranate juice anthocyanins are retained after 400 MPa HHP process. At the end of 21 and 72 days' storage, 65% and 63% of anthocyanins is retained at 4°C (Ferrari et al., 2011). Anthocyanins of blueberry juice samples treated with 600 MPa at 42°C, increased significantly (Barba et al., 2011). No significant change of cyanidin-3-glucoside is observed for HHP treated blood orange juice (Torres et al., 2011). Changes of anthocyanins and total phenolics in fruit mousses and purees after HHP process showed in Table 3. No significant change in concentrations of anthocyanins is also observed for strawberry halves, strawberry and blackberry purees (Terefe et al., 2009; Patras et al., 2009b; Cao et al., 2011). Wild strawberry and strawberry mousses showed 89% retention of anthocyanins after 500 MPa/50°C/10 min. In another study, It is reported that the stability of strawberry and red raspberry anthocyanins, namely pelargonidin-3glucoside and pelargonidin-3-rutinoside, at 800 MPa for 15 min. at moderate temperature (18-22°C) may be due to complete inactivation of polyphenoloxidase (Pozo-Insfran et al., 2006). In contast to these findings, Terefe et al. (2013) and Talcott et al. (2003) showed significant loss of anthocyanins in strawberry puree and muscadine grape juice.

**Table 1.** Changes of anthocyanins and total phenolics in fruits after HHP process

	Process and Storage Conditions	Anthocyanin content		Total Phenolics	References
		After Process	Storage		
Strawberry halves	600 MPa/60°C/10min., 3 months storage at 4°C	No SN effect	27±10% decreased	No SN effect after HHP treatment. 22±13% decreased at the end of storage.	Terefe et al.,2009
Strawberry	200-800 MPa/18- 22°C/15 min., storage at 4-30°C for 9 days		Greater stability at 800 MPa for P3G and P3R at storage temperature of 4°C for 9 days storage.		Zabetakis et al. 2000
Raspberry	200-800 MPa,18-22°C, 15 min., storage at 4- 30°C for 9 days		Greater stability at 800 MPa for C3G and C3S at storage temperature of 4°C for 9 days storage.		Zabetakis et al. 2000
Blackcurrants	200-800 MPa /20- 22,5°C/15 min., storage at 5-30°C for 7 days		Best stability for C3R at 600 MPa 5 °C, for D3R at 800 MPa 5 °C for 7 days storage.		Kouniaki et al. 2004

Anthocyanin content of phenolics generally reduces during storage. While anthocyanins of

HHP-treated cloudy pomegranate, pomegranate and strawberry juices are declining during storage,

anthocyanins of HHP-treated blueberry juices showed no change even after 56 days' storage at 4°C (Chen et al., 2013; Zabetakis et al., 2000; Ferrari et al., 2011, Barba et al., 2011). Significantly higher loss of anthocyanins in HHP-treated cloudy strawberry juices were observed than in HHPtreated clear strawberry juices at 4°C storage, which was possibly due to higher concentrations of oxygen absorbed on pulp particles promoting the degradation of anthocyanins. Moreover, the loss of anthocyanins in both juices at 25°C were significantly higher than at 4°C and only less than 5% anthocyanins were retained in both juices after 6 months of storage at 25°C, which mainly resulted from higher storage temperatures (Cao et al., 2012). Zabetakis al. (2000) also reported 15-20%, 50-60% and 70-80% decreases of anthocyanins in HHP-treated whole strawberry (200-800MPa/15min/18-22°C) after 7 days of storage at 4, 20 and 30°C, respectively. In addition, Ferrari et al. (2011) reported that retention of anthocyanins was %72 and %75 for wild strawberry mousses and strawberry mousses respectively after 21 days' storage, 4°C (Ferrari et al., 2011). The stability of anthocyanins can be due possible complete inactivation polyphenoloxidase as reported for strawberry and red raspberry treated by HHP (Pozo-Insfran et al., 2006). Greater stability of several anthocyanins for raspberry, strawberry, blackcurrants was observed for 600-800 MPa amongst 200-800 MPa treated samples at 4-5°C, 7-9 days' storage (Suthanthangjai et al., 2005; Zabetakis et al., 2000; Kouniaki et al., 2004). With all treatments, it is clear that the extent of anthocyanins losses were lowest when fruits and fruit products were stored at 4°C. Authors mostly reported that anthocyanins are stable to HHP treatment at moderate temperatures, while anthocyanin degradation generally occurs during the storage of the processed foodstuffs. Loss of anthocyanins was probably due to oxidation as well as condensation of anthocyanin pigments with phenolic compounds. Due to the residual oxygen concentration and the activity of polyphenol oxidase and peroxidase, a decreasing trend can be seen during storage. Condensation reactions of anthocyanins with other phenolics naturally occurred in juices with storage times, phenolic acids such as ferulic and syringic acids showed complexities with anthocyanins in strawberry and raspberry juices and condensation products were unstable and further degrade to colorless compounds. The stability of anthocyanins was

also influenced by other fruit components, particularly the interaction of ascorbic acid with anthocyanins. The degredation of anthocyanins induced by ascorbic acid was hypothesized to result indirectly from hydrogen peroxide formation during oxidation of ascorbic acid, which could trigger nucleophilic attack of the C2 of anthocyanins, and finally cause its ring open into check ketone (Castaneda-Ovando et al., 2009; Rein et al., 2005).

### Effects of HHP on total phenolic content (TPC) of fruits and fruit products

The effects of thermal and high hydrostatic pressure (HHP) processing on phenolics of fruits, fruit juices, nectars, purees, mousses, smoothies were assessed by several studies. Total phenolic content (TPC) of pomegranate juice increased significantly after 350 and 500 MPa/30-150 s. HHP process. TPC showed an increase in the first 3 days and started to decrease after 5 days for HHP samples. TPC of control samples was also decreased during storage at 4°C (Varela-Santos et al., 2012). In addition, significant increase of phenolics after 300-400 MPa/2.5-25 treatment was observed in cloudy pomegranate juice (Chen et al., 2013). Alpas (2013) also studied pomegranate juices at 200-400 MPa/5-25°C/5-10 min., no significant change was reported for HHP samples, while significant decrease of phenolics was reported for TT (85°C/10min.).

Changes of anthocyanins and total phenolics in fruit juices, nectars and smoothies after HHP process demonstrated in Table 2. Orange-juice beverage showed significant milk nonsignificant increases for 100-400 MPa HHP treatments and thermal treatments following 90-98°C, 15-21 s. In addition, TPC of smoothie, mix of papaya, melon juices, carrot puree and skimmed milk, after 450-600 MPa HHP treatment increased (Andres et al., 2015). After process, TPC of HHP smoothies was slightly higher (≈11%) compared to TT samples (Keenan et al., 2010; Keenan et al., 2012). Likewise, orange juice milk beverage and smoothies, TPC of prickly pear beverage, apricot nectar and litchi juice increased after HHP process (Jimenez-Aguilar et al., 2015; Huang et al., 2013; Zheng et al., 2014). HHP treatments increased total and individual phenolics in apricot nectars, which were significantly lower than those in HTST-treated apricot nectars (Huang et al., 2013). While HHP increased TPC in litchi juice, a fermented fruit juice, TPC decreased by TT. During 4 weeks of storage at

4°C, TPC in heat treated litchi juice showed a tendency to slowly decrease, while a slight fluctuation is observed in TPC of HHP-treated litchi juice (Zheng et al., 2014). Barba et al. (2011) reported that TPC in HHP-treated and untreated blueberry juice showed no change after process. Likewise, TPC of mango nectar did not change after HHP

(600MPa/1min.) and HTST (110°C/8.6 s). During 56 days' storage of blueberry juices at 4°C, fluctuations in the TPC were observed for all blueberry juice samples. After 16 weeks' storage, TPC decreased by 19.11 and 27.94% in HHP-treated samples, and by 17.03 and 25.23% in HTST-treated samples at 4 and 25°C, respectively.

**Table 2.** Changes of anthocyanins and total phenolics in fruit juices, nectars and smoothies after HHP process

	Process and Storage Conditions	Anthocyanin content		Total Phenolics	
		After Process	Storage		
Pomegranate juice	400 MPa/25°C/5min, 21-72 days storage at 4°C	63% of anthocyanins retained	65% is retained at 21 days storage, 4°C; 63% is retained at 72 days storage, 4°C		Ferrari et al. 2011
Pomegranate juice	350-500MPa/30-150s. 35 days storage	No SN decrease in monomeric anthocyanin pigment concentrations for HHP samples, SN decrease in control samples		TPC is increased significantly between 3.38-11.99% after HHP. TPC showed an increase in the first 3 days and started to decrease after 5 days for HHP samples. The phenolic content is also decreased for control samples during storage.	Varela- Santos et al. 2012
Pomegranate juice	200-400 MPa/5- 25°C/5-10min. or TT (85°C/10min)	No SN decrease in monomeric anthocyanin pigment concentrations for HHP samples, SN decrease in TT samples		No significant decrease in phenolics for HHP samples, significant decrease in TT samples	Alpas 2013
Cloudy pomegranate juice	300-400 Mpa/2.5-25 min. or HTST(110°C/8.6s), 90 days storage at 4°C	Greater retention for HHP samples	Decrease for HHP and HTST samples	SN increase for HHP samples and SN decrease for HTST samples after treatment. During storage, TPC decreased in HHP and HTST samples. There was no significant difference in the total phenols among each detected points from the thirtieth day to ninetieth day during the storage	Chen et al. 2013
Strawberry juice (cloudy and clear)	600 MPa/4 min., 6 month storage at 4- 25°C		29.76% and 7.02% decrease for cloudy and clear samples respectively at 4°C. The decrease at 25°C almost doubled.	16.22% and 13.82% decrease for cloudy and clear samples respectively at the end of 6 months, 4°C.The decrease at 25°C almost doubled.	Cao et al., 2012
Muscadine grape juice	400 -550 MPa /15 min	70 % loss at 400 MPa and 46% loss at 550 MPa			Talcott et al. 2003
Blueberry juice	600 MPa/42°C/5 min. 7-56 days storage	SN increase	No change after 56 days storage	No change in TPC of treated and untreated samples after process.	Barba et al. 2011

				During 56-day storage at 4 °C fluctuations in the TPC were observed for all juices.	
Blood orange juice	400-600 MPa/15 min, storage at 4, 20°C	> %99 retention of C3G is observed	Greater stability at 500 MPa for C3G at 4°C for 10 days storage		Torres et al. 2011
Orange juice- milk beverage	100- 400 MPa for 120-540 s or TT(90- 98°C/15-21 s)			SN and NSN increases for all HHP treatments except 400 Mpa, 540 s SN and NSN increases for all TT	Barba et al. 2012
Litchi juice	500Mpa/2min. or TT (95°C/1min). 4 weeks storage			NSN increase in TPC after HHP treatment, decrease after TT. The content of total phenolics in fermented heat-treated litchi juice showed a tendency to slowly decrease during 4 weeks of storage at 4°C, while a slight fluctuation in the content of total phenolics of fermented HHP-treated litchi juice during 4 weeks of storage at 4°C was observed.	Zheng et al.2014
Prickly pear beverage	400—550 MPa/0.3-16 min.			SN increase in 550MPa/t≥2min. samples	Jimenez- Aguilar et al., 2015
Apricot nectar	300-500MPa/5-20 min. or HTST( 110°C/8.6s)			HHP treatments increased total and individual phenolics in apricot nectars, which were significantly lower than those in HTST-treated apricot nectars.	Huang et al. 2013
Mango nectar	600MPa/1min or HTST(110°C/8.6s) 16 weeks storage at 4 and 25°C, steam blanching was implemented prior both HHP and HTST			No SN changes İN TPC after HHP or HTST.After 16 weeks storage, TPC decreased by 19.11 and 27.94% in HHP-treated samples, and by 17.03 and 25.23% in HTST-treated samples at 4 and 25°C, respectively.	Liu et al. 2014
Fruit smoothie	450-600 MPa/20 °C/1-120 min. or TT(P <sub>70</sub> > 10 min)			TPC of HPP smoothies (450 MPa;5 min) was slightly higher (~11%) compared to TT samples.	Keenan et al. 2010; Keenan et al.2012
Smoothie (mix of papaya, melon juices, carrot puree and skimmed milk)	450-600MPa/20°C/3 min or TT (80°C/3 min.), 45 days storage at 4°C			TPC increased after HHP treatment. TPC decreased 15%, 12%, 11% and 8% for untreated, HHP-450, HHP-600 and TT samples respectively at the end of storage period.	Andrés et al. 2015

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Table 3. Changes of anthocyanins and total phenolics in fruit mousses and purees after HHP process

	Process and Storage Conditions	Anthocyanin content		Total Phenolics	
		After Process	Storage		
Wild strawberry mousses	500MPa/50°C/10min, 21-72 days storage at 4°C	89% of anthocyanins retained	72% is retained at 21 days' storage, 4°C; 63% is retained at 72 days storage, 4°C		Ferrari et al., 2011
Strawberry mousses	500MPa/50°C/10min, 21-72 days storage at 4°C	89% of anthocyanins retained	75% is retained at 21 days' storage, 4°C; 65% is retained at 72 days storage, 4°C		Ferrari et al., 2011
Apple Puree	400-600 MPa/5min/20 °C or TT(75°C/10min.), 7- 21 days storage			No change at 400 MPa, SN decrease at 600 MPa. Storage revealed that mild pasteurization preserved higher levels of phenolics than pressure-treated samples.	Landl et al., 2010
Strawberry and blackberry purees	400-600 MPa/15 min/10-30°C or TT (70°C/2min)	No change at HHP, SN decrease at TT for all samples		Strawberry: no change for 400-500 Mpa, SN increase for 600 MPa, SN decrease for TT Blackberry: no change for 400MPa and TT. SN increase for 500, 600MPa	Patras et al., 2009b
Strawberry puree	600 MPa/20°C/5min. or TT (88°C/2min.), 3 months storage at 4°C	SN decrease for HHP and TT samples	SN decrease for HHP and TT samples	SN decrease for HHP and TT samples after process. Slightly higher loss of TPC were observed during storage of HHP samples compared to TT samples.	Terefe et al., 2013
Strawberry pulp	400-600MPa/5-20min. or TT ((P <sub>70</sub> ≥2 min.)	No SN change for all HHP samples, SN decrease at TT		SN decrease at 400MPa, No SN change at 500, 600MPa, SN increase at TT	Cao et al., 2011
Gooseberry pulp	300-500MPa/3-5min., 60 days storage	nelargonidin-3-rutinosid	e SN: signif	The maximum levels of TPC were observed at 500MPa/5min. at day 0. Some HHP treatments showed reduction in the TPC relate to control samples at day 60.	Vega- Gálvez et al., 2014

C3G: cyanidin-3-glucoside

P3R: pelargonidin-3-rutinoside

C3S: cyanidin-3-sophoroside P3G: pelargonidin-3-glucoside D3R: delphinidin-3-rutinoside

C3R: cyanidin-3-rutinoside

SN: significant

HTST:high temperature short

NSN: nonsignificant TT: thermal treatment TPC:total phenolic content HHP:high hydrostatic pressure

In table 3, changes of anthocyanins and total phenolics in fruit mousses and purees after HHP process is showed. The maximum TPC in gooseberry pulp was observed after 500 MPa/5 min. HHP amongst 300-500 MPa/3-5 min. HHP treatments. Some of HHP-treated samples had higher TPC some of them had lower TPC than control samples (Vega-Gálvez et al., 2014). Similar results were obtained for apricot nectars and apple puree (Huang et al., 2013). TPP content of apple puree was not changed during processing at 400 MPa, but was affected by the 600 MPa and also slightly by the pasteurization (Landl et al., 2010). Based on these results, it can be concluded that HHP treatment, due to changes in fruit pulp microstructure, produces changes in the distribution and aggregation of phenolic compounds. High pressure treatment can increase the rate of mass transfer in an enhancement of solvent penetration into the cells by disrupting the cellular walls and hydrophobic bonds in cell membrane, which may lead to a high permeability (Prasad et al., 2009). Thus, the increase in total phenols may be related to an increased extractability of some antioxidant components such as anthocyanins, amino acids and protein with phenolic hydroxyl group after HHP treatment (Cao et al., 2011).

TPC of strawberry purees and pulps showed different trends after HHP treatments (Patras et al., 2009b; Terefe et al., 2013; Cao et al., 2011). TPC of strawberry purees increased at 600 MPa/10-30°C/15. min., while TPC of strawberry purees showed no change for 400-500 MPa/10-30°C/15 minutes (Patras et al., 2009b). Likewise, Cao et al. (2011) reported no significant change in TPC following HPP (400-600MPa/5-25min.) contrast, Terefe et al. (2013) observed significant decrease at 600MPa/20°C/5min. Terefe et al. (2013) attributed to homogenized strawberry puree samples with substantial tissue disruption, which allowed significant enzyme-substrate interaction during processing due to tissue decompartmentalisation. In addition, strawberry halves, which showed no significant change in TPC after HHP, were vacuum packed limiting the availability of oxygen for oxidation (Terefe et al., 2009). These may explain the observed difference between Terefe et al. (2013) and other studies in the literature like that of Terefe et al. (2009) and Patras et al. (2009b) where the samples were vacuum packed. Patras et al. (2009b) and Terefe et al. (2013) reported that thermal treatment caused significant decrease in TPC of strawberry purees, while Cao et al. (2011)

reported significant increase in TPC of thermally treated strawberry pulps. Cao et al. (2012) reported that TPC of 600 MPa/4 min. HHP-treated cloudy and clear strawberry juices decreased 16.22% and 13.82% respectively at the end of 6 months' storage, 4°C. The decrease at 25°C almost doubled, which was due to decomposition of total phenols induced by higher temperatures during storage (Cao et al., 2012). Normally, polyphenol oxidase and peroxidase were considered to be the main enzymes responsible for the decay of phenols in processed fruits and their derived foods. However, these two enzymes could totally inactivate in blanching, and no enzymatic degradation of total phenols was present in some studies. The decrease of total phenols attributed to the oxidation degradation of phenolic compounds and the polymerization of phenolic compounds with proteins (Cao et al., 2011).

Enzymatic oxidation (polyphenol oxidase and peroxidase) and non-enzymatic autooxidation are responsible of phenolics deterioration. For several PPO enzymes, it has been reported that pressureinduced inactivation proceeds faster at lower pH however the inactivation is also influenced by the addition of salts, sugars or other things. However, it must be noted that the effect of HHP processing parameters such as pressure, temperature and time along with physicochemical properties of fruit such as total soluble solids and pH have varying effects on the enzymes responsible for anthocyanins and phenolics stability in HHP processed fruits and fruit products (Pozo-Insfran et al., 2006).

### **Conclusions and Future Trends**

Phenolic compounds are healthy substances and food industry aims to preserve these compounds and decrease losses of these substances. HHP of fruits and vegetables has been revealed as a useful tool to extend their shelf life and quality as well as to preserve their nutritional and functional characteristics.

Impact of processing conditions and matrix of food on the stability of phenolic compounds in fruits and their products after the process and during storage are very important. It is considered that HHP results in better retention of phenolics and anthocyanins compared to thermal treatments, although some studies indicate that this may not be true in all cases. Processing conditions such as pressure, time, temperature and food-related traits

such as food matrix, pH affects retention of phenolic compounds.

HHP generally proves itself compared to heat treatment in preserving phenolic compounds, but it depends on processing conditions. Further research will be necessary for exploring new applications of HHP technologies, not only to improve the sensorial quality and stability of foods as the main objective, but also to obtain healthy products which have more amounts of bioactive substances like phenolic compounds by preserving them as much as possible during processing. Processing conditions of HHP such as pressure, temperature, time, that increases or stabilizes phenolic compounds, should be studied and determined for more fruit and fruit products. Modelling of HHP conditions for maximum phenolic increase or retainment in different fruits and fruit products would be beneficial.

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